

Synthesis of 6-aryl-2,4-diamino-pyrimidines and triazines using palladium catalysed Suzuki cross-coupling reactions

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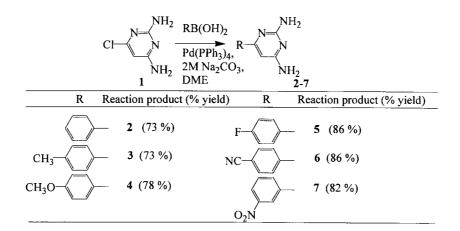
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Abstract—The high yielding synthesis of 6-aryl-2,4-diaminopyrimidines and triazines via palladium catalysed Suzuki cross-coupling reactions of commercially available 6-chloro-2,4-diaminopyrimidine 1 or 6-chloro-2,4-diaminotriazine 8 and aryl boronic acids are described. © 2001 Elsevier Science Ltd. All rights reserved.

1. Introduction

The use of catalytic cross-coupling methodologies for preparing aryl functionalised heterocycles with pharmaceutical, agrochemical, materials and supramolecular applications is a burgeoning field of study.¹ In particular, the so-called Suzuki reaction,² which involves the palladium catalysed cross-coupling of heteroaryl-halides with aryl boronic acids, has received considerable recent attention. We are particularly interested in exploiting the versatility of the Suzuki cross-coupling procedure to prepare 6-aryl-2,4-diamino-pyrimidines and triazines, in view of their potentially interesting molecular recognition,³ agrochemical and medicinal⁴ properties. The synthesis of 6-aryl-2,4diaminopyrimidines are largely unexplored, however, compound **2** has been previously synthesised by condensation reactions of guanidine with acetophenone,⁵ 3-phenylpropynenitrile⁶ or β -bromocinnamonitrile.⁷ The synthesis of 6-aryl-2,4-diaminotriazines have been reported,⁸ and typically involve the condensation of aryl-nitriles or esters with dicyandiamide or biguanide, respectively. Clearly, it would be advantageous to develop cross-coupling strategies utilising Suzuki procedures to allow rapid and convenient synthesis of a range of 6-aryl-2,4-diaminopyrimidines and 6-aryl-2,4-diaminotriazines from a common 6-halo-2,4diamino-pyrimidine or triazine building block.



Scheme 1.

Keywords: triazines; Suzuki cross-coupling; tetrakis(triphenylphosphine)palladium; pyrimidines.

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2. Results and discussion

In the present work, we report the first general, high yielding synthesis of 6-aryl-2,4-diaminopyrimidines 2-7 via Suzuki cross-coupling reactions of commercially available 6-chloro-2,4-diaminopyrimidine **1** with aryl boronic acids (Scheme 1).

We have extended our study to include the first synthesis 6-aryl-2,4-diaminotriazines 9-12 using Suzuki methodology from commercially available 8 and aryl boronic acids, which afforded the triazine derivatives 9-12 in good yield (Scheme 2).

3. Conclusions

In summary, we describe a general, versatile, high yielding method for preparing 6-aryl-2,4-diamino-pyrimidines and an improved method of preparing 6-aryl-2,4-diaminotriazines. Building blocks 1 and 8 will be further exploited to prepare highly functionalised pyrimidine and triazine derivatives, and their synthesis will be reported in due course.

4. Experimental

4.1. General

All reagents were obtained from commercial suppliers and were used without further purification. DME was distilled prior to use. Melting points were measured on a Reichert hot-stage apparatus and are uncorrected. NMR spectra were recorded on a Bruker DPX400 spectrometer. EI Mass spectral data were obtained using a Finnigan MAT 900 XLT mass spectrometer. Elemental analysis data were obtained using an Exeter CE-440 elemental analyser.

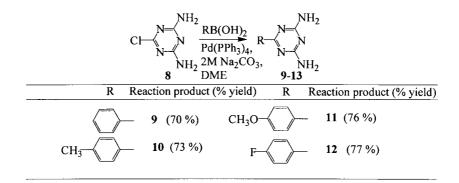
4.1.1. 2,4-Diamino-6-phenylpyrimidine 2. To a mixture 6-chloro-2,4-diaminopyrimidine **1** (0.65 g, 4.5 mmol) and tetrakis(triphenylphosphine)palladium (0.16 g, 0.13 mmol) in DME (10 ml) was added phenylboronic acid (0.823 g, 6.75 mmol) immediately followed by aqueous Na₂CO₃ (2 M, 4.7 ml). The mixture was flushed with N₂ for 5 min and the reaction mixture was then heated under reflux for 48 h. After cooling, the reaction mixture was evaporated under reduced pressure to dryness. THF (100 ml) was

added and the suspension was placed in an ultrasonic bath for a few minutes. The mixture was filtered, washed thoroughly with THF and the filtrate was evaporated under reduced pressure. The residue was purified by column chromatography on silica gel using ethyl acetate as eluent, to afford **2** as a white solid (0.61 g, 73%); $R_{\rm f}$ (ethyl acetate) 0.1; mp 152–156°C (lit.⁵ mp 160–161°C); $\nu_{\rm max}$ (KBr) 3435, 3399, 3335, 3181, 1631; $\delta_{\rm H}$ [400 MHz (CD₃)₂SO]: 7.94 (2H, d, *J*=3.0 Hz, Ph), 7.46 (3H, t, *J*=1.7 Hz, Ph), 6.40 (2H, s, NH₂), 6.26 (1H, s, N=CH), 6.01 (2H, s, NH₂); $\delta_{\rm C}$ [100 MHz (CD₃)₂SO]: 165, 163, 162, 138, 129, 128, 126, 91; MS (EI) 186 (M⁺, 100%). Found: C, 64.62; H, 5.42; N, 30.28. C₁₀H₁₀N₄ requires C, 64.52; H, 5.38; N, 30.11%.

4.1.2. 2,4-Diamino-6-(4-methylphenyl)-pyrimidine 3. This compound was synthesised analogously to **2** using **1** (1.0 g, 7 mmol), 4-methylphenylboronic acid (1.41 g, 10.4 mmol), Na₂CO₃ (2 M, 7.4 ml) and tetrakis(triphenylphosphine)palladium (0.24 g, 0.19 mmol). Compound **3** was obtained as an off-white solid (1.02 g, 73%); $R_{\rm f}$ (ethyl acetate) 0.15; mp 118–120°C; $\nu_{\rm max}$ (KBr) 3435, 3399, 3335, 3181, 1631; $\delta_{\rm H}$ [400 MHz (CD₃)₂SO]: 7.80 (2H, d, J=8.3 Hz, Ph), 7.23 (2H, d, J=7.9 Hz, Ph), 6.33 (2H, s, NH₂), 6.18 (1H, s, N=CH), 5.92 (2H, s, NH₂), 2.33 (3H, s, CH₃); $\delta_{\rm C}$ [100 MHz (CD₃)₂SO]: 165, 163, 162, 140, 135, 129, 126, 90, 21; MS (EI) 200 (M⁺, 100%). Found: C, 65.82; H, 5.98; N, 28.28. C₁₁H₁₂N₄ requires C, 66.00; H, 6.00; N, 28.00%.

4.1.3. 2,4-Diamino-6-(4-methoxyphenyl)-pyrimidine 4. This compound was synthesised analogously to **2** using **1** (1.0 g, 7 mmol), 4-methoxyphenylboronic acid (1.58 g, 10.4 mmol), Na₂CO₃ (2 M, 7.4 ml) and tetrakis(triphenylphosphine)palladium (0.24 g, 0.19 mmol). Compound **4** was obtained as an off-white solid (1.18 g, 78%); $R_{\rm f}$ (ethyl acetate) 0.11; mp 212–215°C; $\nu_{\rm max}$ (KBr) 3435, 3399, 3335, 3181, 1631; $\delta_{\rm H}$ [400 MHz (CD₃)₂SO]: 7.95 (2H, d, J=2.9 Hz, Ph), 6.99 (2H, d, J=2.9 Hz, Ph), 6.30 (2H, s, NH₂), 6.17 (1H, s, N=CH), 5.98 (2H, s, NH₂), 3.76 (3H, s, OCH₃); $\delta_{\rm C}$ [100 MHz (CD₃)₂SO]: 165, 164, 162, 160, 131, 128, 114, 90, 55; MS (EI) 216 (M⁺, 100%). Found: C, 61.22; H, 5.52; N, 26.17. C₁₁H₁₂N₄O requires C, 61.11; H, 5.55; N, 25.93%.

4.1.4. 2,4-Diamino-6-(4-fluorophenyl)-pyrimidine 5. This compound was synthesised analogously to **2** using **1** (0.69 g, 4.76 mmol), 4-fluorophenylboronic acid (1.0 g, 7.14 mmol), Na₂CO₃ (2 M, 5 ml) and tetrakis(triphenyl-phosphine)



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palladium (0.16 g, 0.13 mmol). Compound **5** was obtained as an off-white solid (0.84 g, 86%); $R_{\rm f}$ (ethyl acetate) 0.26; mp 233–236°C; $\nu_{\rm max}$ (KBr) 3435, 3399, 3335, 3188, 2287, 1635; $\delta_{\rm H}$ [400 MHz (CD₃)₂SO]: 7.92 (2H, q, *J*=2.9 Hz, Ph), 7.31 (2H, t, *J*=6.7 Hz, Ph), 7.21 (2H, s, NH₂), 6.83 (2H, s, NH₂), 6.25 (1H, s, N=CH); $\delta_{\rm C}$ [100 MHz (CD₃)₂SO]: 165, 164, 162, 130, 116, 92; MS (EI) 204 (M⁺, 100%). Found: C, 58.90; H, 4.42; N, 27.32. C₁₀H₉N₄F requires C, 58.82; H, 4.41; N, 27.50%.

4.1.5. 2,4-Diamino-6-(4-cyanophenyl)-pyrimidine 6. This compound was synthesised analogously to 2 using 1 (0.69 g, 4-cyanophenylboronic 4.76 mmol). acid (1.05 g, 7.14 mmol), Na₂CO₃ (2 M, 5 ml) and tetrakis(triphenylphosphine)palladium (0.16 g, 0.13 mmol). Compound 6 was obtained as an off-white solid (0.86 g, 86%); $R_{\rm f}$ (ethyl acetate) 0.24; mp 220–224°C (dec). ν_{max} (KBr) 3435, 3399, 3335, 3181, 1631; $\delta_{\rm H}$ [400 MHz (CD₃)₂SO]: 8.05 (2H, d, J=8.0 Hz, Ph), 7.93 (2H, d, J=8.0 Hz, Ph), 6.47 (2H, s NH₂), 6.26 (1H, s, N=CH), 6.06 (2H, s, NH₂); δ_{C} [100 MHz (CD₃)₂SO]: 165, 164, 160, 143, 132, 127, 119, 112, 92; MS (EI) 211 (M⁺, 100%). Found: C, 62.56; H, 4.27; N, 32.28. C₁₁H₉N₅ requires C, 62.56; H, 4.27; N, 33.18%.

4.1.6. 2,4-Diamino-6-(3-nitrophenyl)-pyrimidine 7. This compound was synthesised analogously to **2** using **1** (0.50 g, 3.46 mmol), 3-nitrophenyl boronic acid (0.87 g, 5.19 mmol), Na₂CO₃ (2 M, 3.6 ml) and tetrakis(triphenyl-phosphine)palladium (0.12 g, 0.095 mmol). Compound **7** was obtained as an off-white solid (0.70 g, 82%); $R_{\rm f}$ (ethyl acetate) 0.33; mp 213–216°C; $\nu_{\rm max}$ (KBr) 3435, 3399, 3335, 3181, 1631; $\delta_{\rm H}$ [400 MHz (CD₃)₂SO]: 8.76 (1H, s Ph), 8.30 (2H, d, *J*=1.3 Hz, Ph), 7.72 (1H, t, *J*=8.0 Hz, Ph), 6.51 (2H, s NH₂), 6.34 (1H, s, N=CH), 6.14 (2H, s, NH₂); $\delta_{\rm C}$ [100 MHz (CD₃)₂SO]: 165, 164, 159, 148, 140, 132, 130, 124, 121, 91; MS (EI) 231 (M⁺, 100%). Found: C, 51.72; H, 3.90; N, 30.38. C₁₀H₉N₅O₂ requires C, 51.95; H, 3.90; N, 30.30%.

4.1.7. 2,4-Diamino-6-phenyl-1,3,5-triazine 9. This compound was synthesised analogously to **2** using **8** (1.0 g, 6.87 mmol), phenylboronic acid (1.25 g, 10.3 mmol), tetrakis(triphenyl-phosphine)palladium (0.21 g, 0.19 mmol) and Na₂CO₃ (2 M, 7.2 ml). Compound **9** was obtained as a white solid (0.90 g, 70%); $R_{\rm f}$ (ethyl acetate) 0.1; mp 222–224°C; (_H [400 MHz (CD₃)₂SO]: 8.47 (2H, m, *J*=6.9 Hz, Ph), 7.47 (3H, m, Ph), 3.80 (4H, s, NH₂); MS (EI) 187 (M⁺, 100%). Found: C, 57.62; H, 4.98; N, 37.28. C₉H₉N₅ requires C, 57.75; H, 4.81; N, 37.43%.

4.1.8. 2,4-Diamino-6-(4-methylphenyl)-1,3,5-triazine 10. This compound was synthesised analogously to 2 using 8 (0.25 g, 1.72 mmol), 4-methylphenylboronic acid (0.35 g, 2.58 mmol), tetrakis(triphenylphosphine)palladium (0.09 g, 0.08 mmol) and Na₂CO₃ (2 M, 2 ml). Compound **10** was obtained as a white solid (0.25 g, 73%); $R_{\rm f}$ (ethyl acetate) 0.1; mp 234–235°C; $\delta_{\rm H}$ [400 MHz (CD₃)₂SO]: 8.20 (2H, d,

J=8.3, Ph), 7.20 (2H, d, J=7.9, Ph), 3.70 (4H, s, NH₂), 2.30 (3H, s, CH₃); MS (EI) 201 (M⁺). Found: C, 59.62; H, 5.48; N, 34.78. C₁₀H₁₁N₅ requires C, 59.70; H, 5.47; N, 34.82%.

4.1.9. 2,4-Diamino-6-(4-methoxyphenyl)-1,3,5-triazine 11. This compound was synthesised analogously to **2** using **8** (0.25 g, 1.72 mmol), 4-methoxyphenylboronic acid (0.39 g, 2.58 mmol), tetrakis(triphenylphosphine)palladium (0.09 g, 0.08 mmol) and Na₂CO₃ (2 M, 2 ml). Compound **11** was obtained as a white solid (0.28 g, 76%); $R_{\rm f}$ (ethyl acetate) 0.1; mp>200°C; $\delta_{\rm H}$ [400 MHz (CD₃)₂SO]: 8.42 (2H, d, *J*=9.0 Hz, Ph), 6.99 (2H, d, *J*=9.0 Hz, Ph), 3.90 (3H, s, OCH₃), 3.70 (4H, s, NH₂); MS (EI) 217 (M⁺). Found: C, 55.59; H, 5.00; N, 32.38. C₁₀H₁₁N₅O requires C, 55.30; H, 5.07; N, 32.26%.

4.1.10. 2,4-Diamino-6-(4-fluorophenyl)-1,3,5-triazine 12. This compound was synthesised analogously to **2** using **8** (0.25 g, 1.72 mmol), 4-fluorophenylboronic acid (0.36 g, 2.58 mmol), tetrakis(triphenylphosphine)palladium (0.09 g, 0.08 mmol) and Na₂CO₃ (2 M, 2 ml). Compound **12** was obtained as a white solid (0.27 g, 77%); $R_{\rm f}$ (ethyl acetate) 0.1; mp>200°C; $\delta_{\rm H}$ [400 MHz (CD₃)₂SO]: 8.27 (2H, t, J=9.0 Hz, Ph), 7.28 (2H, t, J=9.0 Hz, Ph), 6.76 (4H, s NH₂); MS (EI) 205 (M⁺). Found: C, 52.62; H, 3.98; N, 34.28. C₉H₈N₅F requires C, 52.68; H, 3.90; N, 34.15%.

Acknowledgements

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References

- 1. Stanforth, S. P. Tetrahedron 1998, 54, 263-303.
- Li, J. J.; Gribble, G. W. *Palladium in Heterocyclic Chemistry*; Tetrahedron Organic Chemistry Series; Pergamon: Amsterdam, 2000; Vol. 20.
- (a) Beijer, F. H.; Kooijiman, H.; Spek, A. L.; Sijbesma, R. P.; Meijer, E. W. Angew. Chem., Int. Ed. 1998, 37, 75–77.
 (b) Deans, R.; Cooke, G.; Rotello, V. M. J. Org. Chem. 1997, 62, 836–839. (c) Beijer, F. H.; Sijbesma, R. P.; Vekemans, J. A. J. M.; Meijer, E. W.; Kooijiman, H.; Spek, A. L. J. Org. Chem. 1996, 61, 6371–6800.
- 4. Pozharskii, A. F.; Soldatenkov, A. T.; Katritzky, A. R. *Heterocycles in Life and Society*; Wiley: Chichester, 1997.
- Wendelin, W.; Zmoelnig, I.; Schramm, H. F. Monatsh. Chem. 1980, 111, 1189–1191.
- Landor, S. R.; Landor, P. D.; Williams, V. E. J. Chem. Soc. Perkin Trans. 1 1984, 2677–2679.
- Iwai, I.; Nakamura, N.; Schnozaki, K. Chem. Abs. 1968, 68, 69035m.
- Smolin, E. M.; Rapoport, L. s-Triazines and Derivatives; Interscience: New York, 1959; Chapter IV, pp 217–268.